

Characterization and Optimization of Mouth Dissolving Film of an Anticoagulant Drug: Apixaban

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ABSTRACT: Oral route is the most convenient route of drug administration among all other routes, but still it is challenging for paediatrics and geriatrics due to swallowing issue. Thus, to overcome this problem MDFs are used. This novel and safer approach; gives rapid systemic action by avoiding first pass metabolism. There are more several advantages like rapid disintegration, no need of water for administration, prevent degradation from acidic environment of stomach. The present research work aimed to prepare mouth dissolving films of an anticoagulant drug apixaban reduces the risk of stroke and systemic embolism in patients with nonvalvular atrial fibrillation. Using Solvent casting method, the films were prepared and optimized using 3² Factorial design considering two independent variables film forming polymer (HPMC E5) and Plasticizer (PEG 400). Dependent variables were taken as disintegration time, drug release and folding endurance. The prepared optimized formulation showed minimum disintegration time, highest dissolution rate (99.8 %) and satisfactory physicochemical properties. From the research, satisfactory results were obtained.

KEY WORDS: Mouth dissolving film, Apixaban, Stroke.

I. **INRODUCTION**

Among, all the routes of drug administration, the most preferred one is oral route because of its convenient, cost effective and ease of

_____ administration that is highly convenient for both paediatrics and geriatrics. Although, it is still a challenging route for swallowing in both paediatrics and geriatrics. So to overcome it, the novel and safer drug delivery such as buccal film, oral strips have been developed. These systems were developed in1970, s as a novel dosage form to overcome the problem of swallowing for both paediatrics and geriatrics and for the systemic drug delivery the film was launched in 2004 [1,25]. The ideal characteristics of that it should be Easy to handle and transport. It should have high stability and ease of administration. It should be easily ionized at oral cavity pH and pleasant in taste. Upto 40% of dose was incorporated in the formulation. It should have high tensile strength and does not stick to packaging material [3]. The advantages of it such as, there is no need of water for administration, accurate dose can be delivered, easy to swallow for both geriatrics and paediatrics, acidic environment of stomach should be avoided. It also gives site specific and local action and provides rapid disintegration and dissolution in oral cavity, due to large surface area. [4, 5]. The disadvantages are that it is not suitable for high dose, the packaging required is expensive, the dose uniformity is a technical challenge and the drugs which are unstable and irritate at buccal pH are not suitable. In addition, restriction of drinking and eating after consumption of oral film for required period of time. [2, 5, 6]



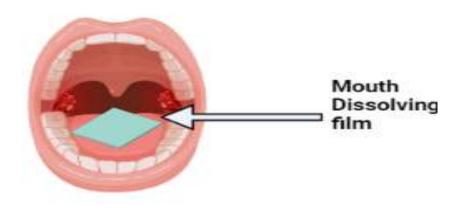


Fig.1 Mouth dissolving film.

Classification of oral film:

Three types of oral films are there: A. Flash release

B. Mucoadhesive melt away wafer

C. Mucoadhesive sustained release Types of oral films with their different properties are summarized in the below table.1

Property	A. Flash Release	B. Mucoadhesive melt	C. Mucoadhesive
		away wafers	sustained release
Area(cm ²)	2-8	2-7	2-4
Thickness(µm)	20-70	50-500	50-250
Structure	Film; single layer	Single or multilayer system	Multilayer system
Excipients	Soluble, highly hydrophilic polymers	Soluble, Hydrophilic polymers	Low/Non soluble polymers
Drug phase	Solid solution	Solid solution or suspended drug particles	Suspension and/or solid solution
Site of action	Systemic or local	Systemic or local	Systemic or local
Dissolution	Maximum 60 seconds	Disintegration in few minutes, forming gel.	Maximum 8-10 hours.
Application	Tongue(upper plate)	Gingival or buccal region	Gingival

Table 1.Types of film and their properties: [7]

II. FORMULATION OF MOUTH DISSOLVING FILM

Active pharmaceutical ingredient:

MDFs can be suitable for various APIs can be for mouth dissolving films. For improving dissolution and uniformity of MDFs; micronized drug can be effective(less than 20mg/day).For drug that is water soluble, there will be no issue of uniformity of distribution. But in water insoluble drug the uniformity may variate, thus to overcome it and for homogeneous distribution for better drug content uniformity, the water insoluble drug is added in milled, micronized form or nanocrystal or microcapsule to get smooth texture of the film. The examples of APIs includes antiasthmatics(e.g., salbutamol),

antiulcer(e.g.,omeprazole),NSAIDS(e.g.,paracetam ol,meloxican,valdecoxib),cough(e.g,dextromethrop han),menstrualpain(ketoprofen), smoking cessation (e.g., nicotine), allergic reaction(e.g., cetrizine, azatadine maleate). [2,8, 9, 10]

Film forming polymer:

For the formation of the film, the polymers play an important role. Hydrophilic polymers can be used for the preparation of various



films. The amount of polymer added should affect the robustness of the film. Minimum 45% w/w of polymer should be present based on total weight of dry film. [60]. Alone as well as in combination the polymer is used, to obtain desired properties of film.

The polymers used must be non irritant, non toxic, and should be inexpensive. It must have

good spreadibility and wetting property. Polymer must have adequate tensile strength. The polymer must have good half life and it does not cause any secondary infection in oral mucosa or in dental site. Both natural and synthetic polymers are used for the preparation [2, 11, and 28].List of such polymers given below in table.2

Natural polymers: Synthetic polymers:	
Pectin	Polyvinyl alcohol
Starch	Hydroxy propyl cellulose
Guar gum	Polyvinyl pyrrolidone
Gelatine	Hydroxy propyl methyl cellulose
Pullulan gum	Sodium carboxy methyl cellulose
Carrageenan gum	Polyethylene oxide
Xanthan gum	Pyroli vinyl pyrollidine

Table 2.Examples	of Natural and	l svnthetic po	lvmers: [3]
		- ~, F ·	

Examples of various polymers with their various properties are described in below table.3

Name of	Molecular	Solubility	Film	pН	Melting
Polymer	weight (g/mol)	·	forming ability	-	point
Hydroxy propyl methyl cellulose	10,000- 1,500,000	Soluble-cold water, Insoluble- chloroform and ethanol	It has film forming ability	5-8	190-200°c
Pullulan	8000- 2,000,000	It is soluble in both hot and cold water	Having high adhesion and film forming capacity	5-7	107°C
Gelatin	15000- 250,000	It is soluble in acid, glycerine and alkali- swell in water	Very good film forming capacity	3.8-6.0	-
Starch and modified starch	50,000- 1,60,000	Insoluble in cold water and ethanol. At 37°C swells in water about 5-10%	Modified starch is having property to form a fast dissolving film	-	250°C
Kollicoat	About 45000	≥50% in water	Good film forming property	6-7	-

 Table 3. List of properties of various films forming polymer:
 [28]

Plasticizer:

Plasticizers should be used to improve the flexibility as well as the mechanical properties of film like tensile strength and elongation and reduce the breakability of the film. A plasticizer selected should be compatible with APIs as well as with the other ingredients. For improving the strip property of plasticizers, the glass transition temperature of polymer for non-aqeous solvent system reduced in the range of 40-60 and for aqeous system the glass transition temperature of polymer is reduced below 75. Examples of some plasticizer are castor oil, polyethylene glycol, citrate derivatives. Etc [2, 12].Various examples of APIs along with the plasticizers used is described in below table.4



Table.4 Examples of APT with plasticizer used: [2, 20, 27, and 28]			
API	Name of Plasticizer		
Triclozan	PEG		
Montelukast sodium	Glycerine		
Sertraline	PEG		
Loperamide	PEG		
Famotidine	PEG		
Ropinirole hydrochloride	PEG		
Cetrizine	PEG		
Telmisartan	PEG		
Dicyclomine hydrochloride	PEG		
Metachlopramide	Glycerol		
hydrochloride			

Table.4 Examples of A	API with plasticizer used: [2, 26, 27, and 28]

Sweetening agents:

Sweetening agents should be used for masking the bitter taste of the APIs. Approximately 3 to 6% w/w concentration of sweeteners should be used in the preparation, either alone or in combination. In the formulation, both natural and artificial sweeteners may be used. Natural sweeteners like sorbitol, mannitol, and isomalt and artificial sweeteners include sucrose neotame, alitame, aspartame, cyclamate may ne incorporate in the films. However, artificial sweeteners are mostly preferable, because natural sugars are restricted for diabetic patients as well as in people who are on diet. [13]

Saliva stimulating agents:

Saliva stimulating agents should be used to increase saliva secretion that helps in faster disintegration of the film. Various acids may be used in the preparation of food can be used as saliva stimulant, such as ascorbic acid, citric acid, lactic acid, tartaric acid and malic acid. Among all the examples the most widely saliva stimulating agent used is citric acid. [9]

Flavouring agents:

Flavouring agents are used to impart flavour to any formulation. Flavouring agents should be compatible with drugs as well as with excipients. Flavours can be extracted from different parts of the plants like flowers, leaves, fruits. Flavours used are natural as well as artificial. Examples are peppermint oil, spearmint oil, cinnamon oil, vanillin, chocolate, apple, pineapple, cherry, and raspberry [8]. Flavouring agents used for masking different taste is descried below in table.5.

 \triangleright

Taste	Flavouring agent used			
Bitter	Mint, anise, walnut, chocolate, wild cherry			
Salty	Peach, butterscotch, vanilla, apricot, maple, winter green			
Sweet	Vanilla, fruit, berry			
Sour	Raspberry, citrus, liquorice root			

Surfactants:

Used as wetting or dispersing or solubilising agent. It is used to dissolve film within seconds and thus immediately release active ingredient. Examples are sodium lauryl sulphate, benzalkonium chloride, tweens, spans, polaxamer 407.Among various examples, mostly Polaxamer 407 is used as wetting, dispersing and solubilising agent [14].Other adjuvant like antioxidants, stabilizers, chelating agent, etc can be used as per need of formulation.

III. MANUFACTURING METHODS

MDFs can be formulated by different methods:-

- Solvent casting method.
- Semisolid casting method.
- \succ Hot melt extrusion.
- Solid dispersion technique.
- ➢ Rolling method.

Solvent casting method.

This method is most commonly used for manufacturing of fast dissolving oral film. In solvent casting method, the water soluble polymers are mixed in water to form homogeneous solution.



Then, the API and remaining excipients are dissolved in smaller amount of other suitable solvent. Both the solutions are combined by stirring and mixing, the air entrapped is removed by sonification. Finally solution is poured in petridish and then dried in the oven. [15]

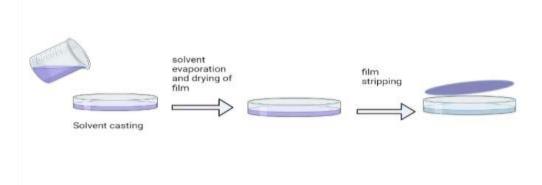


Fig.2 Solvent casting method.

Advantages:

- Films have good physical properties as well as flexibility.
- It is low-cost method.
- It does not cause any changes to API when exposed to high temperature.
- Films have better clarity and gloss.
- Films are free from any damage such as die lines.

Disadvantages:

- The polymer should be dissolved in water or volatile solvent.
- Stable solution should be obtained with moderate viscosity.
- The film formed should be homogeneous.[18]

> Semisolid casting method:

This method is preferred when acid insoluble polymers are used for the preparation of film. Firstly in this method, the solution of water soluble film forming polymer is prepared. The prepared solution is than added to a solution of acid insoluble polymer. Then plasticizer is added in appropriate amount to obtain gel mass. Finally the obtained gel mass is casted in the films or ribbons using heat controlled drums. The ratio of acid insoluble polymer to film forming polymer used should be in 1:4. [16]

Hot melt extrusion method:

This method is mostly used for preparation of granules, transdermal drug delivery system, transmucosal drug delivery system and sustained release tablets. This method includes shaping of polymers through heating. In this method, the drug along with other excipients are combined in dry state, without use of any solvent and then subjected to extruder. Then the extruders having heaters that melt the mixture. The molten mass obtained is shaped in to the films. [17]



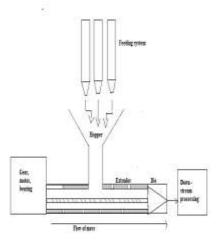


Fig.3 Hot melt extrusion method

Advantages:

- Processing steps are less.
- Solvent or water is not used in this method.
- It is more suitable for poorly soluble drugs.
- Energy needed is less than high shear methods. Disadvantages:
- Thermoliable drugs are not suitable for this method.
- For processing the polymer must have good flow properties.
- It is difficult to maintain dose uniformity.
- Packaging require is expensive.[18]

> Solid dispersion method:

Solid dispersion method is dispersion of one or more active ingredients in an inert carrier in a solid state in the presence of amorphous hydrophilic polymers. Using suitable liquid solvent, the drug is dissolved. Incorporate solution into the melt of polyethylene glycol, below 70°C. Atlast the solid dispersions are shaped into the films by means of dies. [1]

> Rolling method:

In this method, the solvent mainly used are water and mixture of water and alcohol. In small portion of aqueous solvent, the active agent and other ingredients are dissolved by means of high shear processor. Then to prepare homogeneous viscous solution water soluble hydrocolloids is dissolved in water. The solution containing drug is then rolled on a carrier. The films are dried on roller and cutted in desired shape and sizes. [19]

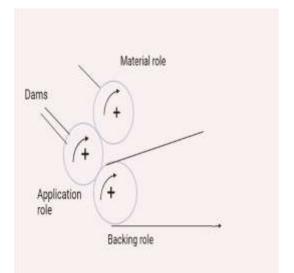


Fig.4 Preparation of MDFs by Rolling method.

IV. METHOD OF PREPARATION Method of Preparation Solvent Casting Method:

The solvent casting method was used for the preparation of the film. Take required amount of drug and then dissolve it in Water. Take required amount of film forming polymer and dissolve in water. Mix both the Solution with continuous stirring and then uniformly dispersed to get clear s solution film forming polymer. After that required amount of plasticizer to be added to film forming solution. Other ingredients and excipients were dissolved one by one in previously prepared film forming polymersolution with constant stirring to form clear solution. The prepared solution were kept in undisturbed condition till the entrapped air bubbles are removed. The aqueous solution was casted in a glass petridish having area of 63.58 cm^2 and was dried at room temperature. The petridish were put on leveled on surface during drying to avoid variation in the thickness. The film took approximately 24 hours to dry at room temperature. The dried film was carefully removed from the mould and was cut into size required for testing. The films were stored in airtight plastic bags till further use.

V. EVALUATION PARAMETERS > Thickness:

The thickness of film strip is measured by using micrometer screw gauge. The thickness of film must not be more than 5% from 5 different locations. The thickness of film is measured to obtain uniformity of film. This step is important



because uniformity in the film thickness is related to the dose accuracy in the film strip. [20, 22]

Folding endurance:

To study the film elasticity, is essential to do folding endurance of film during its handling as well as during its storage. The folding endurance is obtained by cutting a strip of film and continuously folding the film at the same point till it breaks. The number of times, film is folded without breaking considered as folding endurance value.Folding endurance of film is must between 100-150.

> Swelling study:

The film swelling studies is determined using simulated saliva solution. One by one every sample of film is weighed and then put it into a stainless steel mesh that is pre-weighed. In a plastic container containing 15 ml of simulated saliva solution, the mesh containing film samples is merged. At each time interval, the increase in weight of film is measured constant weight is obtained. And percentage elongation is calculated by following equation. [20]

The degree of swelling was calculated by below equation:

w=Wt-Wo/Wo

Wt=It is the weight of film at time t.

Wo=It is the weight if film at time zero.

> Percentage elongation:

When the stress is applied to the sample of film strip, it stretches that is referred as strain. Increase in concentration of plasticizer causes increase in elongation of strip. Percentage elongation of film is calculated by following equation:

Percentage elongation= <u>Increase in length of strip</u> *10

Initial strength of strip

> Tensile strength:

The mechanical strength of the film is measured tensile strength measurement. The point at which maximum stress is applied to break a film is termed as tensile strength. Tensile testing machine like Instron and Monsanto tester is use for testing tensile strength of the film. It can be calculated by the load applied divided by cross sectional area of the film as described in the following equation:

Tensile strength= (load at failure/strip thickness*strip width)*100. [17]

Surface pH:

It is important to determine the surface pH of the film to avoid risk of any side effects by placing film in vivo on the surface of 1.5% w/v agar gel than place pH paper on films. The colour change of pH paper is determined and reported.

The surface pH value for film must be 7 or near to neutral value.

Another method to measure surface pH of film is by using buffer. Cut a film put it in to petridish than add 0.5 ml of buffer solution and measure the surface Ph of film by using Digital pH meter. [1, 21]

Assay/drug content and content uniformity:

For this test standard pharmacopeia is referred, for any particular API, the standard assay method is preferred to determine its content uniformity. Drug content uniformity of the film is measured by UV-Visible spectrophotometer or specify as per pharmacopoeia. In different volumetric flask of 100ml, place films of each formulation and using pH buffer it can be dissolved. The sample of 5ml is withdrawn after 2, 4, 6, 8, 10 and 15 minutes and taken into volumetric flask of 10ml and the volume was made up to the mark. If UV spectroscopy is applied the absorbance is taken, than the value of absorbance is compared against blank in UV spectrophotometer. Using the standard graph, the percentage drug content is determined. [21, 22]

Disintegration time:

For orally disintegrating tablets, the disintegrating time limit is 30 seconds or less that is described in CDER guidance and can be used at development stage or for the quality test as a qualitative guideline. For this study pharmacopeia disintegration test apparatus may be used. Strips have typical disintegration time between 5-30seconds. [23,24]

Percentage drug release:

Ex vivo permeation study was carried out by taking goat oral mucosa using Franz diffusion cell of diameter 2.5 cm. The buccal mucosa than cut and trimmed evenly and then wash it with isotonic solution and immediately keep it between the donor and receptor compartment .Upto 200 ml of buffer is filled in the receptor compartment and temperature maintained at 37±4°C and with contant stirring at 50 rpm was maintained. Than 1 film of 2*2 cm diameter moistened with few drops of buffer solution and placed it in the chamber The donor compartment is filled with 1ml of phosphate buffer. Sample was withdrawn at 2,4, 6, 8, 10 and 15minutes. The amount taken was again filled with fresh buffer solution. The taken sample than diluted and absorbance was taken in UV-visible spectroscopy.

Stability study

Stability study was carried out for 30 days , the optimized formulation was kept at 40 ± 2 °C and 75 ±5 % RH. After completion of 30 days, studies



were carried out for parameters like physical appearance, folding endurance, disintegration time, drug content and % drug release.

VI. RESULT AND CONCLUSION → Preformulation Study Characterization of Drug

So, firstly after receiving the drug we have to check out the basic characteristic of API which is described in below table.

Table 0. Characterization of Drug.				
Test	Result of Analysis			
Description	White crystalline, solid powder			
Solubility	Soluble in water as well as in methanol			
Odour	Odourless			
Melting point	236°C			

Table 6. Characterization of Drug:

Interpretation:

Results are shown in table. API is soluble in water and methanol so solvent casting method can be used for preparation of formulation and the melting point should be in range from 235-238°C described in literature, indicating the purity of drug.

 \rightarrow UV- visible method for determination of Apixaban

Conc.	Absorbance(\lambda max)			Average	S.D
	1	2	3		
1	0.144	0.148	0.148	0.146667	0.002309
2	0.219	0.222	0.227	0.222667	0.004041
3	0.336	0.345	0.355	0.345333	0.009504
4	0.427	0.438	0.467	0.444440	0.020664
5	0.539	0.568	0.574	0.5603330	0.018717
6	0.627	0.641	0.639	0.635667	0.007572
7	0.704	0.719	0.724	0.715667	0.010408
8	0.838	0.840	0.840	0.839333	0.001155
9	0.893	0.898	0.908	0.899667	0.007638

Table 7. UV-	visible method	for identificati	ion of Drug:

Calibration curve of Apixaban was prepared using different concentration from 1, 2, 3, 4, 5, 6, 7, 8, 9 μ g/ml 6.8 phosphate buffer. Average of 3-determinations



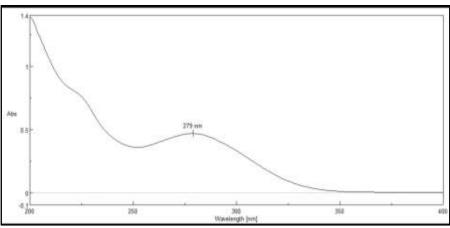


Fig. 5 Scanning of λmax of API

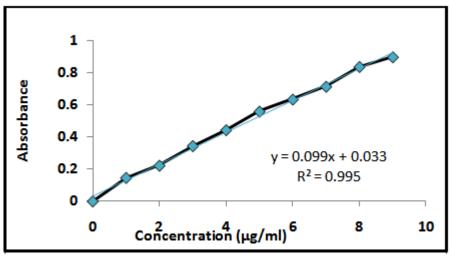
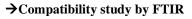
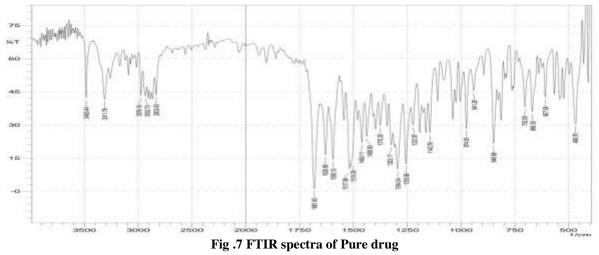


Fig. 6 Calibration curve of Apixaban







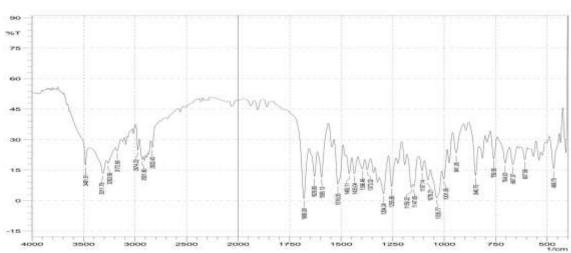


Fig.8 FTIR spectra of Formulation

Stretching	Pure drug (cm ⁻¹)	Formulation(cm ⁻¹)
N-H	3311.78	3311.78
C-H	2976.16	2974.23
=O-NH	1510.28	1516.05
C=0 (Aromatic stretch)	1681.93	1680.00

 Table 8. FTIR data of Apixaban and Optimized Formulation:

Interpretation:

Characteristic peaks obtained for the pure drug correlated well with the polymer as well as the selected formulation peaks. This indicates that the

drug was compatible with the formulation components.

Trial batches of film

Initial trials were taken for finalization of polymers.

Ingredients (mg)	F1	F2	F3	F4	F5	F6
Apixaban	2.5	2.5	2.5	2.5	2.5	2.5
HPMC E3	400	500	-	-	-	-
HPMC E5	-	-	400	500	-	-
HPMC E15	-	-	-	-	400	500
PEG 400(ml)	0.5	1.5	0.5	1.5	0.5	1.5
Citric acid	20	20	20	20	20	20
Beta cyclodextrin	2.5	2.5	2.5	2.5	2.5	2.5

Table 9.	Composition	of Mouth	dissolving film:
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Water	q.s	q.s	q.s	q.s	q.s.	q.s.

The prepared films of trial batches were evaluated for basic evaluation parameters. The results were recorded below in tabulated form.

Formul ation Code	Stickiness	Surface appearan ce	Film clarity	Quality	Foldi ng endu rance	Disintegr ation time (sec)	рН	Weight variation (mg)
F1	Non sticky	Non uniform	Clear	Average	150± 20	40±11	6.8±0.3	133±2.7
F2	Non stocky	Non uniform	Turbid	Average	200± 12	38±17	7.1±0.2	135±4.4
F3	Non sticky	Uniform	Clear	Good	204± 10	33±12	7.1±0.1	130±2.3
F4	Non sticky	Uniform	Clear	Good	248± 19	35±15	7.0±0.2	127±3.5
F5	Non sticky	Non- Uniform	Turbid	Average	210± 15	129±17	6.92±0. 1	129±4.1
F6	Non sticky	Non- Uniform	Turbid	Average	252± 20	148±10	7.0±0.3	139±1.7

Table 10.	Evaluation a	of Trial batches:	

Interpretation:

From the above, F1 and F2 are from HPMC E3, F3 and F4 are from HPMC E5 polymers and F5 and F6 from HPMC E15. HPMC E3 and E15 make average films, while HPMC E5 makes good films.

Formulation Batches (3² Factorial Design)

After Trial batches, HPMC E5 was selected and further by applying 3² factorial designs total 9 formulations were suggested and prepared using solvent casting method.



R (L NO	Table 11. 3 ² Facto	orial Designs:	X2		
Batch NO.		X1 Amount of HPMC E5		of PEG 400	
A1	0		-1		
A2	0		0		
A3	+1		-1		
A4	-1		+1		
A5	0		+1	+1	
A6	-1	-1			
A7	+1	+1			
A8	+1		0	0	
A9	-1		0		
Independent Variables	Level				
-	Low (-1)	Medium	(0)	High (+1)	
HPMC E5 (X1) mg	400	450		500	
PEG 400 (X2) ml	0.5				
Dependent Variables	·	•		•	
Response Y1 : Folding endura	ince				
Response Y2 : Disintegration	time				
Response Y3 : % Drug release					

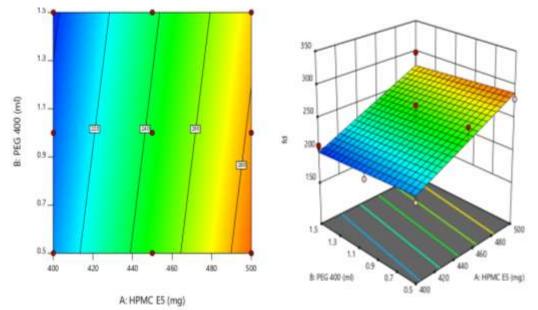
Table 11 22 Fastarial Desi

All batches were evaluated for Folding endurance, Disintegration time and % Drug release to find the effect of both X1 and X2 on the film

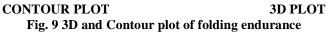
1	able 12. rac	torial Date	ches of A	pixaban I	Mouth D	issoiving	гшп;		
Ingredients(mg)	A1	A2	A3	A4	A5	A6	A7	A8	A9
Apixaban	2.5	2.5	2.5	2.5	2.5	2.5	2.5	2.5	2.5
HPMC E5	450	450	500	450	400	400	500	500	400
PEG 400 (ml)	0.5	1	0.5	1.5	1.5	0.5	1.5	1	1
Citric acid	20	20	20	20	20	20	20	20	20
Sodium Starch Glyconate	30	30	30	30	30	30	30	30	30
Beta cyclodextrin	2.5	2.5	2.5	2.5	2.5	2.5	2.5	2.5	2.5
Water	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s

Table 12 Factorial Batches of Anixaban Mouth Dissolving Film.





> Folding Endurance Contour plot and 3D plot:

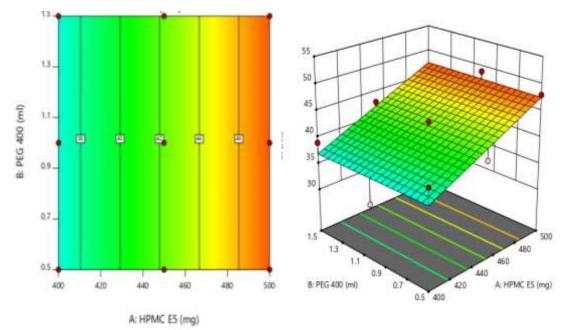


Source	Sum of Squares	df	Mean Square	F - value	p- value	
Model	9486.83	2	4743.42	8.54	0.0176	significant
A-HPMC E5	9282.67	1	9298.67	16.17	0.0064	
B- PEG 400	204.17	1	204.17	0.3674	0.5666	
Residual	3334.06	6	555.68			
Cor Total	12820.89	8				

Table 13. ANOVA	for Response 1: Foldin	g endurance

For folding endurance, the ANOVA table have significant impact on film.





> Disintegration time Contour plot and 3D plot:

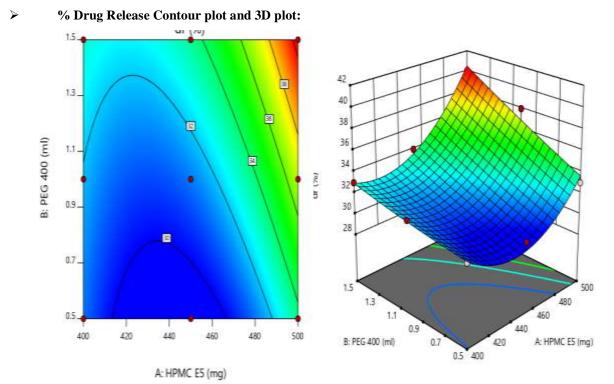
CONTOUR PLOT 3D PLOT Fig. 10 3D and Contour plot of Disintegration time

Source	Sum of Squares	df	Mean Square	F - value	p- value	
Model	170.67	2	85.33	10.47	0.0110	significant
A-HPMC E5	170.67	1	170.67	20.95	0.0038	
B- PEG 400	0.0000	1	0.0000	0.0000	1.0000	
Residual	48.89	6	8.15			
Cor Total	219.56	8				

Table 14. ANOVA for Response 2 : Disintegration time

For Disintegration time ,, the ANOVA table have significant impact on film.





CONTOUR	PLOT		3D PLOT
Fig. 11	3D and	Contour plot of %	6 Drug release

Table 15. ANOVA for Response 3 : %Drug release						
Source	Sum of Squares	df	Mean Square	F - value	p- value	
Model	92.03	5	18.14	13.16	0.0297	significant
A-HPMC E5	37.50	1	37.50	26.82	0.0140	
B- PEG 400	28.17	1	28.17	20.15	0.00206	
Residual	4.19	3	1.40			
Cor Total	96.22	8				

Fable 15. ANOVA	for Response 3 :	%Drug release
	for hesponse e	/ of a groupe

For Percentage drug release the ANOVA table have significant impact on film

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Table 16. Evaluation of Mouth Dissolving film				
Weight variation(mg)	Swelling index(mg)	Surface Ph	Tensile Strength (N/mm ²)	
<u>133±3.5</u> 128±2.5	237 <u>±1.8</u> 240 <u>+</u> 1.4	6.9 <u>+</u> 0.2 7.0 <u>+</u> 0.1	3.33±0.05 4.52±0.02	
131 <u>+</u> 3.4	240 <u>+</u> 2.5	6.9 <u>+</u> 0.3	3.60±0.02	
130 <u>+</u> 1.2	257 <u>+</u> 3.1	6.8 <u>+</u> 0.3	3.42±0.03	
124 <u>+</u> 2.9	234 <u>+</u> 2.2	6.9 <u>+</u> 0.2	4.16±0.04	
125 <u>+</u> 4.1	238 <u>+</u> 1.4	6.9 <u>+</u> 0.3	4.62±0.02	
130 <u>+</u> 2.2	260 <u>+</u> 1.1	6.9 <u>+</u> 0.2	3.47±0.04	
138 <u>+</u> 2.5	266 <u>+</u> 1.2	7.2 <u>+</u> 0.1	3.78±0.05	
152 <u>+</u> 3.4	230 <u>+</u> 2.4	7.0 <u>+</u> 0.4	3.49±0.02	
	Weight variation(mg) 133 ± 3.5 128 ± 2.5 121 ± 3.4 130 ± 1.2 124 ± 2.9 125 ± 4.1 130 ± 2.2 138 ± 2.5	Weight variation(mg)Swelling index(mg) 133 ± 3.5 128 ± 2.5 237 ± 1.8 240 ± 1.4 131 ± 3.4 131 ± 3.4 240 ± 2.5 130 ± 1.2 124 ± 2.9 257 ± 3.1 124 ± 2.9 125 ± 4.1 234 ± 2.2 130 ± 2.2 130 ± 2.2 260 ± 1.1 138 ± 2.5 266 ± 1.2	Weight variation(mg)Swelling index(mg)Surface Ph 133 ± 3.5 128 ± 2.5 237 ± 1.8 240 ± 1.4 6.9 ± 0.2 7.0 ± 0.1 131 ± 3.4 240 ± 2.5 240 ± 2.5 6.9 ± 0.3 130 ± 1.2 257 ± 3.1 234 ± 2.2 6.8 ± 0.3 124 ± 2.9 234 ± 2.2 234 ± 2.2 6.9 ± 0.2 125 ± 4.1 238 ± 1.4 260 ± 1.1 6.9 ± 0.2 130 ± 2.2 260 ± 1.1 260 ± 1.1 6.9 ± 0.2 138 ± 2.5 266 ± 1.2 7.2 ± 0.1	

Table 16 Evolution of Mouth Dissolving fil

Table 17. Evaluation of Mouth dissolving film :

Formulation	% Elongation	Thickness(mm)	Folding endurance	Disintegration time(sec)	Drug content (%)
A1	30.12±3.19	0.36 <u>+</u> 0.02	208 <u>+</u> 19	<u>39+</u> 10	98.9+1.2
A2	39.44±4.12	0.37 <u>+</u> 0.03	270 <u>+</u> 16	40 <u>+</u> 12	99.7+1.0
A3	26.66±5.16	0.30 <u>+</u> 0.04	298 <u>+</u> 20	46 <u>+1</u> 8	97.2+2.2
A4	38.54±6.28	0.36 <u>+</u> 0.03	194 <u>+</u> 13	32 <u>+</u> 8	85.5+3.1
A5	47.11±5.64	0.32 <u>+</u> 0.02	279 <u>+</u> 14	48 <u>+</u> 13	98.9+1.3
A6	40.22±6.47	0.34 <u>+</u> 0.05	261 <u>+</u> 22	49 <u>+2</u> 0	98.6+1.1



A7	38.66±3.33	0.37 <u>+</u> 0.05	269 <u>+</u> 11	<u>40+</u> 8	88.9+2.1
A8	42.18±2.23	0.38 <u>+</u> 0.03	207 <u>+</u> 16	43 <u>+1</u> 5	99.5+2.2
A9	45.19±4.77	0.34 <u>+</u> 0.02	200 <u>+</u> 14	40 <u>+</u> 12	99.7+1.8

→Percentage Drug Release:

Table 18. % Drug Release :

Formulation	Time (min)		<u></u>			
	2	4	6	8	10	15
A1	33.4	50.3	62.4	77.0	88.4	98.6
A2	30.6	52.4	68.2	76.2	85 .4	99.8
A3	31.2	45.5	53.3	67.2	78.5	85.7
A4	32.4	45.6	58.3	79.1	89.6	99.7
A5	48.6	52.4	60.4	73.3	84.7	96.6
A6	49.5	55.8	61.5	73.4	84.8	97.8
A7	40.3	42.7	57.8	68.5	79.9	83.9
A8	43.4	48.8	56.0	70.3	77.1	82.2
A9	40.1	53.3	64.1	75.6	83.1	94.8



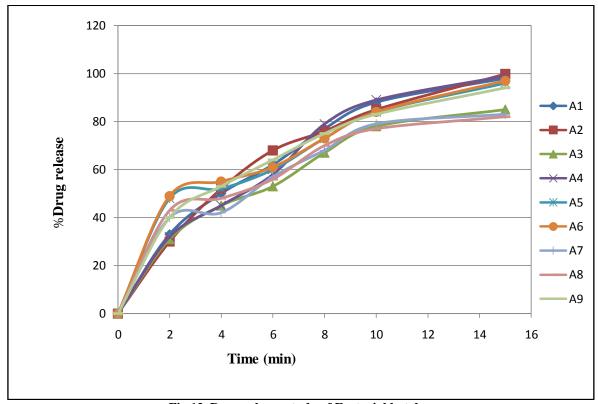


Fig 12. Drug release study of Factorial batches

Interpretation :

A2 was showing the good dissolution →Rate Kinetics Study

Table 19. R ² value of Rate kinetics study.
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FORMULATIONS	ZERO ORDER	FIRST ORDER	HIGUCHI MODEL	KORSMEYER PEPAS
	R ²	R ²	R ²	R^2
A1	0.9282	0.8489	0.9952	0.9878
A2	0.95974	0.8895	0.9917	0.9925
A3	0.9408	0.8762	0.9711	0.9836
A4	0.9217	0.8615	0.9638	0.9757
A5	0.9681	0.9503	0.9567	0.9829
A6	0.9786	0.9638	0.9639	0.9357
A7	0.8961	0.863	0.928	0.913
A8	0.9112	0.8876	0.9426	0.9375
A9	0.9475	0.8864	0.9901	0.9986

Kinetic modeling data of Optimized batch A2



Table 20. Rate kinetics data of Optimized batch A2					
Kinetic Model	Parameters	Value			
Zero Order	R ²	0.95974			
First Order	R ²	0.8895			
Korsmeyer-Peppas	R ²	0.9925			
Higuchi Model	R ²	0.9917			

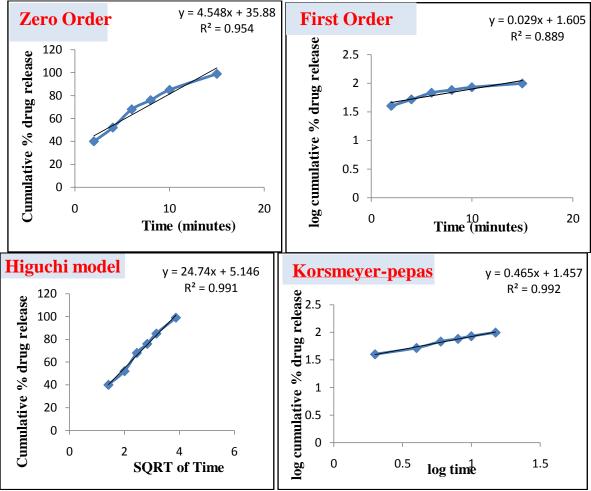


Fig 13. Rate kinetics study of Optimized batch

Interpretation :

The drug release data of batch A2 is fitted in to different kinetic models. Among all, the best fitted model is Korsmeyer-Peppas model because R² value of Korsmeyer-Peppas model has 0.9925



Table : 25 Result of stability study of optimized batch A2				
Paramters	Initial	After 1 month		
Appearance	Clear and Transparent	Clear and transparent		
Folding endurance	270	262		
Drug content	99.7	99.4		
% Drug release	99.8	99.6		
Disintegration time (sec)	40	45		

→ STABILITY STUDY

INTERPRETATION:

The A2 batch stability study was found to be satisfactory.

VII. CONCLUSION:

The aim of the present study was to optimize and characterize mouth dissolving film of Apixaban. Firstly the trial batches of film were prepared using 3 different polymers HPMC E3, HPMC E5 and HPMC E15 and from the result HPMC E5 is selected. Preformulation was performed to investigate the drug-excipient compatibility by doing FTIR of 2 samples that is 1) Pure Drug, 2) Final formulation and the spectra was found to be satisfactory. Further, in UV, the graph obtained was Linear. Then, the formulation was optimized based on 3² factorial design in that, PEG 400 and HPMCE5 was taken as independent variable. Analysis of factorial design done for dependent variables such as Folding endurance, Disintegration time and % Drug release. All the results obtained was found to be satisfactory. Among all, the final formulation A2 was selected based on result obtained by evaluation. Also the F2 formulation shows good Weight variation, Surface pH, Tensile strength, % Thickness, Elongation, Disintegration with in 40 seconds, Drug release is 99.8% in 15 minutes, Drug content is 99.7 and Folding endurance 270.

REFERENCE

 Rathod, S., Phansekar, M., Bhagwan, A., & Surve, G. (2013). A review on mouth dissolving tablets. Indian Drugs, 50(11), 5– 14. https://doi.org/10.53879/id.50.11.p0005

- [2]. Sharma, D., Kaur, D., Verma, S., Singh, D., Singh, M., & Singh, G. (2015). Fast dissolving oral films technology: A recent trend for an innovative oral drug delivery system. International Journal of Drug Delivery, 7(2), 60–75. https://doi.org/10.5138/ijdd.v7i2.1692
- [3]. Gupta, M., Gupta, R., Khunteta, A., & Swarnkar, S. (2017). An Overview of Mouth Dissolving Films: Formulation Aspects. International Journal of Pharmaceutical and Biological Science Archive, 5(5), 01–18. www.ijpba.in
- [4]. Pawar, R., Sharma, R., Sharma, P., & Darwhekar, G. N. (2019). A Review on Mouth Dissolving Film. Journal of Drug Delivery and Therapeutics, 9(6), 206–210. https://doi.org/10.22270/jddt.v9i6.3676
- [5]. Kahsu, A., Aklilu, T., Masresh, B., & Melkam, W. (2015). Benefits and Risks of Fluoroquinolones Use in Pediatrics: a Review. International Journal of Life Sciences and Review Review Article International Journal of Life Sciences and Review, 1(5), 169–174. https://doi.org/10.13040/IJPSR.0975-8232.IJLSR.1
- [6]. Bhattarai, M., & Gupta, A. K. (2016). Fast Dissolving Oral Films: A Novel Trend to Oral Drug Delivery System. Sunsari Technical College Journal, 2(1), 58–68. https://doi.org/10.3126/stcj.v2i1.14802
- [7]. MS, A., & C, V. (2016). Formulation and Evaluation of Fast Dissolving Oral Films of Diazepam. Journal of Pharmacovigilance,



4(3), 25–40. https://doi.org/10.4172/2329-6887.1000210

- [8]. Ghodake, P. P., Karande, K. M., Osmani, R. A., Bhosale, R. R., & Harkare, B. R. (2013). Mouth Dissolving Films: Innovative Vehicle for Oral Drug Delivery Mouth Dissolving Films: Innovative Vehicle for Oral Drug Delivery. September.
- [9]. Vishwakarma, D. K., & Banerjeer, A. (2018). Acta Biomedica Scientia mouth dissolving film: an approach to novel drug delivery system-a review. 5(1), 29–36.
- [10]. Dahiya, M., Saha, S., & Shahiwala, A. F. (2009). A Review on Mouth Dissolving Films. 469–476.
- [11]. Rajeev, G. (2019). An Advanced Oral Solid Dosage Form : Oral Dissolving Film. 9, 731–742.
- [12]. Meghana, R., & Velraj, M. (2018). An overview on mouth dissolving film. 11(4), 12–15.
- [13]. Bhyan, B., Jangra, S., Kaur, M., & Singh, H. (2015). Review Article Orally Fast Dissolving Films: innovations in formulation and technology. 9(2), 50–57.
- [14]. Sharma P., Sharma P.,Darwhekar G.,Shrivastava,(2018). Review Article 1 An overview about novel fast dissolving oral films. 6(1), 1–7
- [15]. Arya, A., Chandra, A., Sharma, V., & Pathak, K. (2010). Fast Dissolving Oral Films : An InnovativeDrug Delivery System and Dosage Form. 2(1), 576–583.
- [16]. R., Kumar, S. N., Kumar, K. R., Sundaram, S. R., Sudharsan, S., & Nadu, T. (2021). An overview on mouth dissolving films. 10(2), 656–673.
- https://doi.org/10.20959/wjpps20212-18213
- [17]. Panda, B. (2016). Fast Dissolving Film as an Effective Orally Disintegrating Drug Delivery System Development of Innovative Orally Fast Disintegrating Film Dosage Forms : A Review. July.
- [18]. Reddy, M. R. (2020). An Introduction to Fast Dissolving Oral Thin Film Drug Delivery Systems : A Review. 12(7), 925– 940.
- [19]. Reza, K. H., & Chakraborty, P. (2016). Recent industrial development in Oral Thin Film Technology: An Overview. 4(8), 17– 22.
- [20]. Patil, P., & Shrivastava, S. K. (2014). Fast Dissolving Oral Films : An Innovative Drug Delivery System. 3(7), 2088–2093.

- [21]. R, M. D., Puja, C., & Sridhar, K. A. (2016). Asian Journal of Research in Chemistry and. April.
- [22]. Subash V., Basani G., 1, Guru S, Rao M.,Overview on fast dissolving film. (2010).
- [23]. I., Naik, T. S., Khale, A., & Kanekar, H. (2014). International Journal of Pharmaceutical and Evaluation of Mouth Dissolving Films : Physical and Chemical Methods. 4(1), 62–65.
- [24]. Jyoti, A., Gurpreet, S., Seema, S., & Rana, A. C. (2011). ISSN 2230 – 8407 Review Article fast dissolving films: a novel approach to oral drug delivery. 2(12), 69–74.
- [25]. Tiwari, R. R., Umashankar, M. S., & Damodharan, N. (2018). Recent update on oral films: A bench to market potential. International Journal of Applied Pharmaceutics, 10(6), 27–33. https://doi.org/10.22159/ijap.2018v10i6.287 25
- [26]. Alagusundaram, M. (2011). Development and evaluation of novel-transbuccoadhesive fi lms of Famotidine. 2(1), 17–23. https://doi.org/10.4103/2231-4040.79800
- [27]. Sandeep, I. A., Bhavana, N., Sankeerth, P. T., Nandhini, P., & Nadendla, R. R. (2017). FORMULATION AND EVALUATION OF LOPERAMIDE ORAL FILMS. 8(10), 370–379.
- [28]. [28]. Thakur, N., Bansal, M., Sharma, N., Yadav, G., & Khare, P. (2013). Overview "A Novel Approach of Fast Dissolving Films and Their Patients." Advances in Biological Research, 7(2), 50–58. https://doi.org/10.5829/idosi.abr.2013.7.2.72 134
- [29]. [29]. https://en.wikipedia.org/wiki/Polyethylene_g lycol
- [30]. Arjun, M., & Joshi, L. (2019). "FORMULATION AND EVALUATION OF SUBLINGUAL F ILMS OF APIXABAN." International Journal of Pharmaceutical and Biological Science Archive PubMed, 7(2). www.ijpba.in